

Trishomocubanone (Pentacyclo[6,3,0,0^{2,6},0^{3,10},0^{5,9}]undecan-4-one)

By PHILIP E. EATON,* RICHARD A. HUDSON, and CLAUDIO GIORDANO
(Searle Chemistry Laboratory, The University of Chicago, Chicago, Illinois 60637)

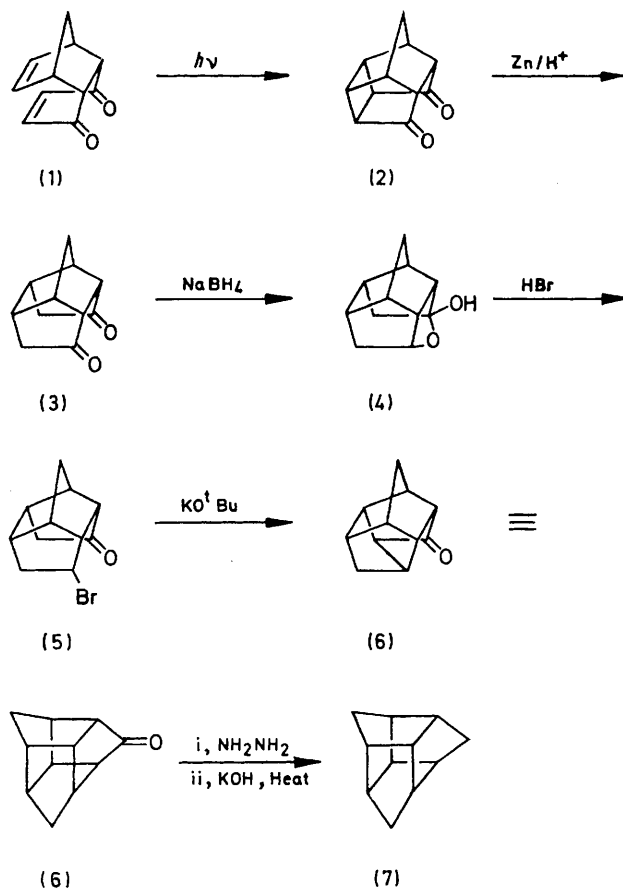
Summary A method is outlined for the conversion of the cyclopentadiene-quinone Diels-Alder adduct into trishomocubanone and trishomocubane.

WE have found a direct synthetic route to trishomocubanone (6).¹ The pentacyclic framework of this ketone, unlike that in most other highly condensed cage systems, contains no cyclobutane rings: the skeleton is constructed of five-membered rings. It is intrinsically chiral. Trishomocubanone is a key member of the system as the carbonyl group provides a ready 'handle' for optical resolution and for the introduction of other functional groups.

Unlike other entries into the trishomocubane system our synthesis does not involve a carbonium ion rearrangement.^{1b,c} The well known cage compound (2), produced by photocyclization of the cyclopentadiene-*p*-benzoquinone Diels-Alder adduct (1),² is reduced with Zn-AcOH to the tetracycle (3).^{1a,3} Reduction of (3) with NaBH₄ in aqueous EtOH gives, in nearly quantitative yield, the hydroxyketone as the internal hemiacetal (4), m.p. 223–223.5° (from hexane). Treatment of (4) with excess of 32% HBr in AcOH at 100° for 8 h under pressure gives at least 64% of the bromoketone (5), m.p. 63–65° (from hexane), ν_{\max} (CCl₄) 1748 cm⁻¹. This bromoketone reacts quantitatively at room temperature with a suspension of KO^tBu in anhydrous Et₂O to produce the title ketone (6), m.p. 163–164° after sublimation, ν_{\max} (CCl₄) 1785 (sh), 1778, and 1757 cm⁻¹, 2,4-dinitrophenylhydrazone m.p. 207.5–208.5 (from MeCN). Wolff-Kishner reduction of (6) using standard Huang-Minlon conditions gives the hydrocarbon (7), m.p. 149–151° (from MeCN) (lit.^{1b} 147–149°).

The ¹H and ¹³C n.m.r. spectra of (6) and (7) are completely consistent with the structural assignments. Compound (6), ¹H n.m.r. (270 MHz; CDCl₃) δ 2.47 (2H, *W*₁ ca. 12 Hz), 2.40 (4H, *W*₁ ca. 7 Hz), 1.79 (2H, *W*₁ ca. 10 Hz), 1.69 (2H, d, *J* 10.5 Hz), and 1.43 (2H, d, *J* 10.5 Hz); ¹³C n.m.r. (22.63 MHz; proton-decoupled; CDCl₃) δ (from Me₄Si) 217.3(w), 50.2, 47.6, 41.1, 40.9, and 35.5 p.p.m. (intensity of upper 5 lines approximately equal). Compound (7), ¹H n.m.r. (270 MHz; CDCl₃) δ 2.0 (6H, *W*₁ ca. 10 Hz), 1.8 (2H, *W*₁ ca. 15 Hz), and 1.32 (6H, *W*₁ ca. 3 Hz); ¹³C n.m.r. (22.63 MHz; proton-decoupled; CDCl₃) δ 47.7, 41.5, and 33.2 p.p.m., intensity ratio roughly 6:2:3. The presence of only three resonances in the proton-decoupled ¹³C n.m.r. spectrum of (7) establishes unambiguously the arrangement of the methano-bridges in this trishomocubane as shown. Other permutations would give rise to five or more resonances in the ¹³C spectrum. (There are altogether nine

possible arrangements of the methano bridges, excluding mirror images).



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¹ (a) Entry into this ring system was reported first by one of us at the 155th Meeting of the American Chemical Society, San Francisco, April, 1968. The method presented then was a primitive version of that described here. (b) Another approach to the ring system (but not the ketone) has been given by G. R. Underwood and B. Ramamoorthy (*Tetrahedron Letters*, 1970, 4125). (c) P. v. R. Schleyer, S. A. Godleski, E. Osawa, and G. J. Kent, preceding communication, have synthesised trishomocubane by AlBr₃-isomerization of the hydrocarbon derived from (2). We thank Dr. Schleyer for informing us of this result prior to publication.

² R. C. Cookson, E. Crundwell, and J. Hudec, *Chem. and Ind.*, 1958, 1003.

³ E. Wenkert and J. E. Yoder, *J. Org. Chem.*, 1970, 35, 2986.